

Unusual Boron Trifluoride-catalyzed Reactions of Taxinine Derivatives with α - and β -4(20)-Epoxydes

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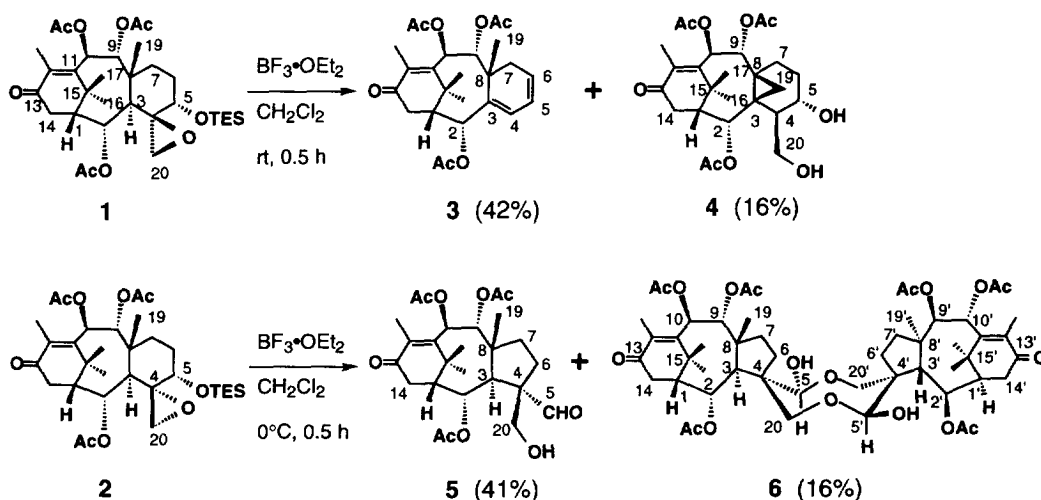
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Abstract: Boron trifluoride-catalyzed reaction of β -4(20)-epoxy-5-*O*-triethylsilyltaxinine A (**1**) gave the 3,5-diene (**3**) and the 3,8-cyclopropane (**4**) derivatives, while the similar reaction of the corresponding α -4(20)-epoxide (**2**) afforded the ring contracted derivative (**5**) and its hemiacetal dimer (**6**). Plausible mechanisms of these reactions containing 1,2-hydride shift (**3** and **4** from **1**) or pinacol-type rearrangement (**5** and **6** from **2**) are proposed.

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In our continuing chemical derivatization from taxinine,^{1,2} one of major taxoids obtained from the Japanese yew *Taxus cuspidata*,^{3–5} we found that boron trifluoride-catalyzed reaction of β -4(20)-epoxy-5-*O*-triethylsilyl (TES) taxinine A (**1**) yielded two unexpected compounds, the 3,5-diene (**3**) and the 3,8-cyclopropane (**4**) derivatives, while the similar reaction of the corresponding α -4(20)-epoxide (**2**) afforded the ring contracted derivative (**5**) and its hemiacetal dimer (**6**). In this paper we describe the formation of **3** and **4** from **1** and that of **5** and **6** from **2**, and propose plausible mechanisms of these reactions.



Scheme 1

Treatment of the β -4(20)-epoxy-5-*O*-TES-taxinine A (**1**), which was derived from taxinine A,¹ with boron trifluoride diethyl etherate (BF₃•OEt₂, 2eq.) in CH₂Cl₂ at room temperature for 0.5 h yielded the 3,5-diene (**3**, 42%) and the 3,8-cyclopropane (**4**, 16%) (Scheme 1), while the reaction at 0°C afforded the 3,5-diene (**3**, 47%) but the 3,8-cyclopropane (**4**) was not obtained. When the reaction mixture was treated with 2,4-dinitrophenylhydrazine, formaldehyde-2,4-dinitrophenylhydrazone was obtained, suggesting the liberation of formaldehyde in this reaction. On the other hand, treatment of the α -4(20)-epoxide (**2**), which was also derived from taxinine A,¹ with BF₃•OEt₂ (2 eq.) in CH₂Cl₂ at 0°C for 0.5 h afforded the cyclopentanecarbaldehyde (**5**, 41%) and its hemiacetal dimer (**6**, 16%) (Scheme 1). The structures of **3** ~ **6** were elucidated by spectral data including MS and 2D NMR as follows.⁶

Compound **3** was shown to have the molecular formula, C₂₅H₃₂O₇, by HREIMS [*m/z* 444.2150 (M⁺), Δ +0.1 mmu]. The ¹H NMR spectrum of **3** showed three olefine proton signals at δ_{H} 5.58 (H-6), 5.68 (H-4), and 5.75 (H-5). The three olefinic methine carbons (δ_{C} 119.5, C-4; 122.4, C-5; 123.5, C-6) were indicated by the HMQC spectrum. ¹H-¹H COSY connectivities of H-4 to H-5, H-5 to H-6, and H-6 to H₂-7 and HMBC correlations of H-2 to C-3 and C-4 revealed that **3** possessed a conjugated diene (C-3 ~ C-6) in ring C, which was supported by UV absorption at 258 nm (ϵ 6800).

The molecular formula, C₂₆H₃₆O₉, of compound **4** was established by HREIMS [*m/z* 492.2383 (M⁺), Δ +2.3 mmu]. The ¹H and ¹³C NMR data (δ_{H} 0.43 and 0.77, d, *J* = 5.4 Hz, H₂-19; δ_{C} 21.0, t, C-19, 27.6, s, C-8, and 28.2, s, C-3) of **4** indicated the presence of a cyclopropane ring.⁷ The cyclopropane ring was fused between C-3 and C-8 on the basis of HMBC correlations of H-2 to C-19, H-4 to C-3 and C-19, H-7 to C-19, and H-9 to C-8. A hydroxymethylene group (δ_{H} 2.83 and 3.58, H₂-20; δ_{C} 65.9, t, C-20) was connected at C-4 by an HMBC correlation between H-5 and C-20. NOESY correlations of H-19a and H₃-16 to H-2 and H-9 indicated that the cyclopropane ring was β -oriented, while β -orientation of the hydroxymethylene group at C-4 was elucidated by NOESY correlations of H-19b to H-20a and H-14a to H-4.

Compound **5** was shown to have the molecular formula, C₂₆H₃₆O₉, by HRFABMS [*m/z* 493.2452 (M+H)⁺, Δ +1.4 mmu]. The ¹H and ¹³C NMR and HMQC spectra of **5** showed one formyl (δ_{H} 9.37, s, H-5; δ_{C} 202.8, d, C-5) and one hydroxymethylene (δ_{H} 4.19, 3.60, d, *J* = 11.4 Hz, H₂-20; δ_{C} 62.6, t, C-20) groups. HMBC correlations of H-3 and H₂-7 to C-4 (δ_{C} 61.9, s) and H₃-19 to C-3, C-7, and C-8 indicated the presence of a cyclopentane ring (ring C), while the formyl and hydroxymethylene groups were attached at C-4 on the basis of HMBC correlations of H-5 to C-3 and C-20 and H₂-20 to C-6. The β -orientation of the hydroxymethylene group at C-4 was elucidated by NOESY correlations of H₃-19 to H-20a and H-5 to H-3 and H-14b. Thus compound **5** was shown to possess a 6/8/5-ring system with a formyl and a hydroxymethylene groups at C-4.

The molecular formula, C₅₂H₇₂O₁₈, of compound **6** was established by HRFABMS [*m/z* 985.4812 (M+H)⁺, Δ +1.5 mmu], indicating that **6** was a dimer of **5**. The ¹H and ¹³C NMR data of **6** were very similar to those of **5**, except for resonances of a formyl and a hydroxymethylene groups at C-4 in **5**. The presence of two hemiacetal moieties (δ_{H} 5.18, brs, H-5; δ_{C} 99.2, d, C-5; δ_{H} 4.79, brs, H-5'; δ_{C} 105.7, d, C-5') was revealed by the HMQC spectrum. HMBC correlations of H-3 and H-5 to C-4, H-5' to C-20, and H-3' to C-5', C-4', and C-20' and a NOESY correlation between H-5 and H-5' indicated the presence of a 1,5-dioxacyclooctane moiety. NOESY correlations of H-5 to H-20a and H-5', H-5' to H-20a and H-

20'a revealed that the 1,5-dioxacyclooctane ring had a chair/chair conformation and the two hydroxy groups at C-5 and C-5' were equatorial. Thus compound **6** was assigned as a dimer of **5**.

The formation of the 3,5-diene (**3**) and the 3,8-cyclopropane (**4**) from the β -4(20)-epoxide (**1**) may be explained by 1,2-hydride shift and then liberation of formaldehyde or cyclopropanation, respectively (Figure 1). In the first step $\text{BF}_3 \cdot \text{OEt}_2$ induces the fission of C-4 - O bond followed by 1,2-shift of hydride from C-3 to C-4, resulting in generation of the carbocation at C-3 (**a**). Liberation of formaldehyde and dehydration of the allylic functional group at C-5 in **a** give the 3,5-diene (**3**). On the other hand, the 3,8-cyclopropane (**4**) can arise from a trapping of the carbocation center formed at C-3 (**a**) by the methyl (Me-19) followed by loss of a proton.^{8,9} In the case of the α -4(20)-epoxide (**2**) $\text{BF}_3 \cdot \text{OEt}_2$ induces the fission of C-4 - O bond followed by ring contraction through a pinacol-type rearrangement (**b**) to afford the cyclopentanecarbaldehyde (**5**). The pinacol-type rearrangement involving deprotection of the TES group at C-5 accompanies with transfer of the C-6 carbon into C-4 position and generation of an α -oriented formyl group. It is noted that the orientation of the 4(20)-epoxides results in different types of the boron trifluoride - catalyzed reactions.

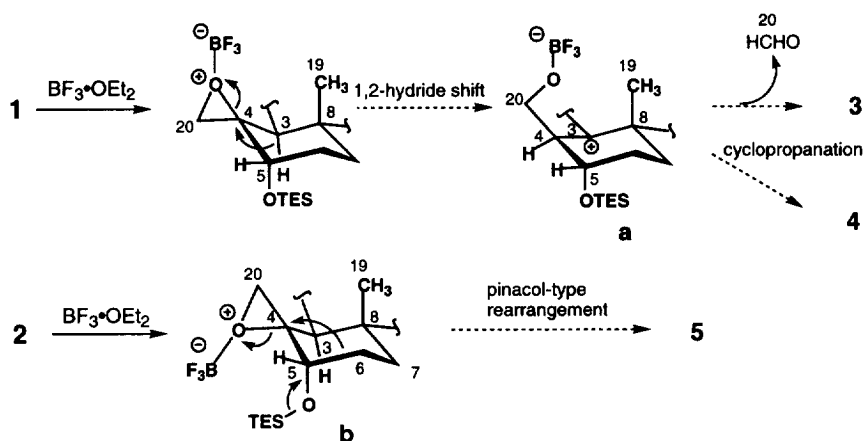


Figure 1. Plausible mechanisms of $\text{BF}_3 \cdot \text{OEt}_2$ -catalyzed reactions of compounds **1** and **2**

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6. Compound **1** (34 mg) was dissolved in dry CH₂Cl₂ (0.4 mL), and then added BF₃•OEt₂ (46% Et₂O solution, 40 µl) under Argon. The solution was stirred at room temperature for 0.5 h, and the reaction mixture was extracted with CHCl₃ and saturated aqueous NaHCO₃. The organic layer was washed with water and brine, dried over Na₂SO₄, and evaporated. The residue was applied to a silica gel column (*n*-Hexane/EtOAc, 4:1) to afford compounds **3** (10.6 mg) and **4** (4.4 mg). According to the same procedure as described above, compounds **5** (3.4 mg) and **6** (1.3 mg), which were separated by a silica gel HPLC (Develosil 60-5, Nomura Chemical; CHCl₃/EtOAc, 3:1), were obtained from **2** (10 mg). **Compound 3**: [α]_D²⁵ +233° (c 1.00, CHCl₃); IR (film) ν_{max} 1745, 1674, 1237, 1020 cm⁻¹; UV (MeOH) λ_{max} 258 nm (ε 6800); ¹H NMR (CDCl₃): δ_H 6.02 (1H, d, *J* = 10.2 Hz), 5.97 (1H, d, *J* = 10.2 Hz), 5.75 (1H, m), 5.68 (1H, d, *J* = 6.0 Hz), 5.66 (1H, d, *J* = 2.7 Hz), 5.58 (1H, m), 2.74 (1H, dd, *J* = 19.8, 6.0 Hz), 2.69 (1H, m), 2.61 (1H, d, *J* = 19.8 Hz), 2.28 (1H, m), 2.27 (1H, dd, *J* = 18.7, 6.2 Hz), 2.09 (3H, s), 2.06 (3H, s), 2.04 (3H, s), 1.96 (3H, s), 1.73 (3H, s), 1.17 (3H, s), 1.05 (3H, s); ¹³C NMR (CDCl₃): δ_C 198.7, 170.2, 170.1, 169.0, 148.0, 140.3, 139.6, 123.5, 122.4, 119.5, 75.7, 74.6, 69.8, 49.0, 42.0, 37.3, 36.1, 34.2, 31.0, 25.7, 23.6, 21.5, 21.5, 21.0, 14.1. **Compound 4**: [α]_D²⁴ +39.1° (c 1.00, CHCl₃); IR (film) ν_{max} 3447, 1741, 1655, 1237, 1021 cm⁻¹; UV (MeOH) λ_{max} 263 nm (ε 2000); ¹H NMR (C₆D₆): δ_H 6.41 (1H, d, *J* = 10.3 Hz), 5.86 (1H, d, *J* = 10.3 Hz), 5.41 (1H, d, *J* = 1.6 Hz), 3.60 (1H, m), 3.58 (1H, dd, *J* = 10.3, 2.5 Hz), 3.36 (1H, d, *J* = 19.9 Hz), 2.83 (1H, m), 2.80 (1H, dd, *J* = 19.9, 7.6 Hz), 2.60 (3H, s), 2.40 (1H, m), 2.13 (1H, dd, *J* = 7.6, 1.6 Hz), 1.93 (1H, ddd, *J* = 13.0, 12.0, 4.9 Hz), 1.87 (3H, s), 1.79 (3H, s), 1.70 (3H, s), 1.67 (1H, m), 1.67 (3H, s), 1.32 (1H, m), 1.26 (1H, m), 0.99 (3H, s), 0.77 (1H, d, *J* = 5.4 Hz), 0.43 (1H, d, *J* = 5.4 Hz); ¹³C NMR (C₆D₆): δ_C 198.0, 169.5, 169.0, 168.9, 152.1, 139.0, 78.7, 77.9, 74.7, 65.9, 64.3, 51.4, 44.6, 38.1, 37.2, 36.0, 28.2, 27.6, 25.9, 23.4, 21.0, 20.7, 20.4, 20.3, 16.8, 15.2. **Compound 5**: [α]_D²⁵ +79.7° (c 1.00, CHCl₃); IR (film) ν_{max} 3489, 1744, 1676, 1233, 1025 cm⁻¹; UV (MeOH) λ_{max} 264 nm (ε 5400); ¹H NMR (CDCl₃): δ_H 9.37 (1H, s), 6.05 (1H, d, *J* = 9.4 Hz), 5.87 (1H, d, *J* = 9.4 Hz), 5.46 (1H, dd, *J* = 7.7, 3.7 Hz), 4.19 (1H, d, *J* = 11.4 Hz), 3.60 (1H, d, *J* = 11.4 Hz), 3.27 (1H, d, *J* = 7.7 Hz), 2.71 (1H, dd, *J* = 20.0, 7.5 Hz), 2.31 (1H, d, *J* = 20.0 Hz), 2.26 (1H, dd, *J* = 7.7, 7.5 Hz), 2.26 (1H, m), 2.12 (3H, s), 2.08 (3H, s), 2.06 (3H, s), 2.04 (3H, s), 1.81 (1H, m), 1.68 (3H, s), 1.59 (2H, m), 1.13 (3H, s), 1.11 (3H, s); ¹³C NMR (CDCl₃): δ_C 202.8, 199.3, 170.5, 169.7, 169.0, 150.2, 138.1, 77.3, 73.8, 71.0, 62.6, 61.9, 49.5, 47.7, 46.7, 37.8, 37.6, 35.7, 34.7, 26.0, 25.4, 21.0 (2C), 20.9, 20.6, 13.8. **Compound 6**: [α]_D²⁶ +51.1° (c 1.00, CHCl₃); IR (film) ν_{max} 3481, 1748, 1673, 1235, 1083 cm⁻¹; UV (MeOH) λ_{max} 266 nm (ε 8700); ¹H NMR (CDCl₃): δ_H 6.38 (1H, d, *J* = 9.9 Hz), 6.32 (1H, d, *J* = 9.9 Hz), 6.17 (1H, d, *J* = 9.9 Hz), 6.13 (1H, d, *J* = 9.9 Hz), 5.83 (1H, dd, *J* = 6.5, 2.8 Hz), 5.82 (1H, dd, *J* = 6.8, 3.1 Hz), 5.18 (1H, brs), 4.79 (1H, brs), 4.12 (1H, d, *J* = 12.0 Hz), 4.02 (1H, d, *J* = 11.8 Hz), 3.81 (1H, d, *J* = 11.8 Hz), 3.73 (1H, brd, *J* = 12.0 Hz), 3.43 (1H, d, *J* = 6.5 Hz), 3.27 (1H, d, *J* = 6.8 Hz), 2.83 (1H, dd, *J* = 19.8, 7.2 Hz), 2.60 (1H, m), 2.59 (1H, dd, *J* = 19.4, 7.1 Hz), 2.51 (1H, d, *J* = 19.8 Hz), 2.50 (3H, s), 2.34 (3H, s), 2.338 (3H, s), 2.32 (1H, m), 2.29 (1H, d, *J* = 19.4 Hz), 2.26 (1H, dd, *J* = 7.2, 2.8 Hz), 2.16 (1H, dd, *J* = 14.4, 9.3 Hz), 2.11 (1H, dd, *J* = 7.1, 3.0 Hz), 1.83 (1H, m), 1.82 (3H, s), 1.81 (3H, s), 1.76 (3H, s), 1.74 (3H, s), 1.72 (2H, m), 1.70 (3H, s), 1.69 (3H, s), 1.68 (3H, s), 1.67 (1H, m), 1.64 (1H, m), 1.25 (3H, s), 1.11 (3H, s), 0.95 (6H, s); ¹³C NMR (CDCl₃): δ_C 200.3, 198.1, 169.9, 169.7, 169.2, 169.1, 168.6, 168.4, 151.3, 151.1, 138.4, 138.2, 105.7, 99.2, 76.4, 76.2, 74.1 (2C), 73.7, 71.6, 71.5, 66.2, 54.7, 51.9, 49.9, 49.8, 47.3 (2C), 46.4, 45.2, 38.2, 38.0, 37.6, 37.5, 36.5, 35.9, 34.2, 33.9, 27.4, 25.1 (2C), 23.1, 21.5, 21.2, 20.3 (2C), 20.2 (2C), 19.3, 13.6, 12.9.
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